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Grid-Type Two-Dimensional Magnetic Multinuclear Metal Complex: Strands of $\{[Cu^{II}(\mu-4,4'-bpy)]^{2+}\}_n$ Cross-Linked by Octacyanotungstate(V) lons

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Reaction of the preorganized strands of $\{ [Cu^{\parallel}(\mu-4,4'-bpy)]^{2+} \}_n (4,4'-bpy) \}_n$ bpy = 4,4'-bipyridine) with $[W^{V}(CN)_{8}]^{3-}$ leads to a novel cyanobridged $Cu^{II}_{3}W^{V}_{2}$ complex $[Cu(\mu-4,4'-bpy)(DMF)_{2}][Cu(\mu-4,4'-bpy) (DMF)]_{2}[W^{V}(CN)_{8}]_{2} \cdot 2DMF \cdot 2H_{2}O 1$. The structure of 1 consists of the expected 2-dimensional grid-type network which is built of infinite $\{[Cu^{\parallel}(\mu-4,4'-bpy)]^{2+}\}_n$ chains cross-linked by octacyanotungstate units. The Cu^{II}-NC-W^V-CN-Cu^{II} linkage exhibits the topology of a 3,2-chain. The skeleton of the layer is additionally stabilized by a hydrogen bond network formed by terminal cyano ligands of the $[W(CN)_8]^{3-}$ moiety and water molecules. The distance between the adjacent Cu₃^{II}W₂^V chains within the layer is 11.12 Å along the *a* axis. The layers are connected by H-bonds of NCN-NDMF-NCN linkages into 3-D supramolecular architecture. The magnetic properties correspond to a dominant ferromagnetic coupling within the Cu^{II}₃W^V₂ pentamer units (J = +35(4) cm⁻¹) and much weaker effective AF interunit coupling which include both intra- and inter-3,2-chain interactions between pentamers (J' $= -0.05(1) \text{ cm}^{-1}$).

The application of octacyanometalate $[M(CN)_8]^{n-}$ (M = Mo, W, and Nb) and 3d transition metal molecular precursors in design and construction of new crystalline materials¹ with potentially adjustable magnetic properties^{1b,e,g,j-1,2} has developed recently into an area of intense research. In the design of new solid state architectures, the strategy of self-assembly through the formation of coordinate M–CN–M' bonds between mononuclear molecular precursors accompanied by weaker $\pi - \pi$ stacking, hydrogen bonding, and electrostatic interactions has been widely employed. A logical extension of the building-block approach is to combine the preorganized multinuclear supramolecular coordination compound

formed by 3d metal centers bridged by organic *spacer* ligands with an octacyanometalate moiety. In order to explore this idea we have chosen the predictable coordination polymer network of 1-dimensional chains that are generated from the metal ions coordinated to linear bifunctional ligands such as 4,4'-bipyridine (4,4'-bpy).³ These polymeric strands are usually cross-linked by hydrogen-bonded bridging 4,4'-bpy ligands. The new strategy consists of the cross-linking of

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 ${[M^{II}(\mu-4,4'-bpy)]^{2+}}_n$ chains by $[M(CN)_8]^{3-/4-}$ spacers. Herein, we report the synthesis and magnetic properties of $[Cu(\mu-4,4'-bpy)(DMF)_2][Cu(\mu-4,4'-bpy)(DMF)]_2[W^V(CN)_8]_2$ • 2DMF•2H₂O **1**.

Compound 1 was obtained in 62% yield in a two step process. In the first step the dropwise addition of a solution of Cu(ClO₄)₂·6H₂O (37.1 mg, 0.1 mmol, 10 mL) in MeOH/ MeCN 2:1 to a stirred solution of 4,4'-bipyridine (15.6 mg, 0.1 mmol, 10 mL) in MeOH/MeCN 2:1 gave a violet suspension. DMF (12 mL) was added to dissolve the suspension, resulting in a pale-blue solution. In the second step the resulting solution was slowly added to a 50 mL solution of (NBu₄)₃[W^V(CN)₈] (111.9 mg, 0.1 mmol) in MeCN. The small amount of greenish-blue precipitate immediately formed was filtered off, and the filtrate was subject to crystallization by the slow evaporation method. After 1 week light blue needle-shaped X-ray quality crystals were formed in a very low yield precluding CHN analysis. The composition of 1 was established by X-ray analysis. The $\nu_{\rm CN}$ IR spectrum of 1^4 shows sharp absorptions at 2188m, 2171m, and 2143sh, consistent with maintaining the W^V site and the presence of bridging cyano groups.^{1f}

X-ray analysis⁵ revealed that the compound consists of the expected 2-dimensional sheet structure which is built of infinite { $[Cu^{II}(\mu-4,4'-bpy)]^{2+}_n$ chains cross-linked by octacyanotungstate units (Figure 1a). The Cu^{II}-NC-W^V-CN-Cu^{II} linkage thus formed exhibits the topology of a 3,2chain^{1h,6} assured by the electroneutrality of **1** (Figure 1b). Relevant bond distances and angles are given in Table 1. The W^V site exhibits a distorted square antiprism coordination geometry with metric parameters typical for a [W(CN)₈]³⁻ building unit.^{1f} Three of the cyano groups of [W(CN)₈]³⁻ are linked to Cu^{II} sites. [Cu(3)(μ -4,4'-bpy)(DMF)(NC)₂] moieties of compressed square pyramidal geometry occupy the alternate corners of Cu₂^{II}W₂^V squares, whereas [Cu(2)-(μ -4,4'-bpy)(DMF)₂(NC)₂] of elongated octahedral geometry acts as a linker between the two squares in the chain. The

(4) IR spectrum of 1: v(O−H) 3511s, 3452s, 3381sh; v(ArC−H) 3145w, 3104w, 3084w, 3058w, 3018w; v(C−H) 2962sh, 2928s; v(H−C=O) 2873m, 2856m, 2814w; v(C=N) 2188s, 2171s, 2143sh; v(ArC=C), v(C=O) 1649vs, 1610vs, 1532m, 1489m, 1433s, 1414s; v(C−C, C−N), v(O−H, C−C, C−O, ArC−H in-plane) 1386s, 1366sh, 1298sh, 1247m, 1217m, 1148w, 1118w, 1099m, 1070m, 1052sh, 1012w, 977sh, 953sh; v(ArC−H out-of-plane) 866w, 814s, 728m 688m, 660sh, 646sh; v(W−C, Cu−N) 573w, 521w, 483m, 454m.

(5) Crystal and refinement parameters: $1 \text{ C}_{64}\text{H}_{70}\text{Cu}_3\text{N}_{28}\text{O}_8\text{W}_2$, T = 295 K, $P2_12_12$, Z = 2, a = 17.6167(1) Å, b = 19.0794(3) Å, c = 11.1204-(4) Å, V = 3737.75(15) Å³, $d_{calc} = 1.704$ g/cm³, R1 = 0.0273, wR2 = 0.0822. Data were collected on a Nonius KappaCCD diffractometer using graphite-monochromated Mo K α ($\lambda = 0.71073$ Å) radiation and were corrected for Lorentz, polarization, and absorption effects. The structure was solved by direct methods using SIR-92 (Altomare et al.) and refined against all data using SHELXL-97 (Sheldrick). The crystal of 1 exhibited racemic twinning described by the twin matrix -10 00-10/0 0 -1, and the twin component factor was refined to 0.146(9). The positions of H atoms (except those of water molecules) were calculated using geometrical constraints and included in the refinement using riding model.



Figure 1. The crystal structure of $[Cu(\mu-4,4'-bpy)(DMF)_2][Cu(\mu-4,4'-bpy)(DMF)]_2[W^V(CN)_8]_2 · 2DMF · 2H_2O. Atoms not involved in bridging are omitted for clarity. (a) Side view of 2-D layer. (b) 3-D H-bonding network of$ **1**in [001] direction with relevant atomic numbering scheme.

Table 1. Relevant Bond Lengths (Å) and Angles (deg) for 1

[W(C	N)8] ³⁻
W1-C range/average	2.162(7)-2.192(7)/2.173(11)
C-N range/average	1.105(9)-1.140(8)/1.131(11)
W1-C-N range/average	174.4(5)-179.7(6)/177.7(1.6)
Cu2(bpy) ₂ (I	$NC)_2(DMF)_2$
Cu2-N12	2.007(5)
Cu2-N12-C12	151.1(6)
Cu3(bpy) ₂ (NC) ₂ (DMF)
Cu3-N13	2.297(5)
Cu3-N13-C13	174.3(5)
Cu3-N15	1.992(5)
Cu3-N15-C15	177.9(5)

structure of the layer is additionally stabilized by a hydrogen bond network formed by terminal cyano ligands of the $[W(CN)_8]^{3-}$ moiety and water molecules. The distance between the adjacent $Cu^{II}_3W^{V_2}$ chains within the layer is 11.12 Å along the *a* axis. The layers are connected by H-bonds of N_{CN}-N_{DMF} -N_{CN} linkages into 3-D supramolecular architecture.

Magnetic measurements were performed using a Quantum Design SQUID magnetometer. The temperature dependence of the $\chi_M T$ product is shown in Figure 2. The $\chi_M T$ equals 2.0 cm³ mol⁻¹K at 300 K and gradually increases with decreasing temperature, reaching a maximum of 3.52 cm³ mol⁻¹ K at 5 K. Zero field cooled and field cooled susceptibility measured at 20 Oe (not shown) does not show any trace of irreversibility nor anomaly down to 2 K, which points to lack of a 3-D magnetic ordering. The field

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Figure 2. Temperature dependence of $\chi_M T$ value for 1 in the external magnetic field of 1 kOe. Inset: Field dependence of magnetization for 1 at 2 K. Solid lines represent calculated fits to the data.

dependence of the magnetization measured at 2 K reaches 4.85 $\mu_{\rm B}$ per formula unit at 50 kOe, which is very close to the value expected for five saturated $\frac{1}{2}$ spins with g = 2.0 $(5.0 \mu_{\rm B})$. In 1, there are two copper sites, Cu(2) and Cu(3), and three different [W-CN-Cu] linkages (Scheme S1). The $[Cu(2)(bpy)_2(NC)_2(DMF)_2]$ moiety exhibits symmetrical equatorial cyano bridges in the [W-CN-Cu(2)] linkage. The [Cu(3)(bpy)₂(NC)₂(DMF)] moiety reveals unsymmetrical equatorial and axial cyano bridges in [W-C15-N15-Cu-(3)] and [W-C13-N13-Cu(3)] linkages, respectively. In the equatorial cyano bridging at Cu(2,3) centers the magnetic orbital of Cu $(3d_{x^2-y^2})$ of σ -symmetry) is orthogonal to the magnetic orbital of W (of π -symmetry, centered on the 5d_{z²} partially delocalized on the π^* orbitals of the cyanide), so that the magnetic interaction is expected to be ferromagnetic (Scheme S2). In contrast, the axial cyano bridging at the Cu(3) center gives rise to the π overlap of the magnetic orbitals of Cu and W, leading to the antiferromagnetic interaction. The magnetic coupling in the $Cu^{II}(4,4'-bpy)$ -Cu^{II} linkage is expected to be weak and antiferromagnetic.⁷ On the basis of this qualitative considerations, the magnetic system of 1 can be treated as Cu^{II}₃W^V₂ pentamer units with identical Cu^{II}····W^V ferromagnetic intrapentamer interactions defined by the parameter J and much weaker effective AF interunit coupling (J'), which includes both intra- and inter-3,2-chain interactions between pentamers (Scheme 1). The Hamiltonian $\mathbf{H}_5 = -J(S_{Cu1}S_{W1} + S_{W1}S_{Cu2} + S_{Cu2}S_{W2} +$ $S_{W2}S_{Cu3} - g\beta H(S_{Cu1} + S_{W1} + S_{Cu2} + S_{W2} + S_{Cu3})_z$, which represents a chain of five 1/2 spins with the isotropic Heisenberg exchange interaction, was solved numerically. Molar magnetization of the pentamer was calculated as M_5 = $N_A \sum_n (-\partial E_n / \partial H) \exp(-E_n / k_B T) / \sum_n \exp(-E_n / k_B T)$, where E_n denotes eigenvalues of \mathbf{H}_5 . The low temperature limit of $(\chi_5 T)_{\text{LT}}$ (where $\chi_5 = M_5/H$) reaches 4.35 cm³ mol⁻¹ K in this model, which is significantly higher than the experimental value. The difference and the presence of the maximum in the $\chi_{\rm M} T(T)$ curve was attributed to AF interactions between pentamer units through the cyano- and 4,4'-

Scheme 1. The Exchange Interactions in **1** Represented by J (Solid Lines) and J' (Dotted Lines)



bpy bridges (J') and taken into account in the mean field model. At low H fields, where the M(H) dependence is linear, the molar susceptibility was calculated using $\chi_M = \chi_5/(1 - \lambda\chi_5)$ relation where $\lambda = 4J'/N_Ag^2\beta^2$. To calculate M(H) at high fields the mean field equation $M = M_5(H + \lambda M)$ was solved numerically. The best agreement with the data was obtained for $J = +35(4) \text{ cm}^{-1}$, $J' = -0.05(1) \text{ cm}^{-1}$, and fixed g = 2. This set of parameters satisfactorily reproduces the $\chi_M T$ product throughout the entire temperature range, and also the magnetization curve at 2 K. The J' value compares favorably with the value estimated for magnetic coupling in the Cu^{II}-(4,4'-bpy)-Cu^{II} linkage ($J_{\text{eff}} = -0.052 \text{ cm}^{-1}$).⁷

In summary, a new assembly based on one-dimensional coordination polymer { $[Cu(\mu-4,4'-bpy)]^{2+}_n$ and octacyanometalate(V) was designed and constructed: $[Cu(\mu-4,4'-bpy)-(DMF)_2][Cu(\mu-4,4'-bpy)(DMF)]_2$ [$W^V(CN)_8]_2 \cdot 2DMF \cdot 2H_2O$ **1** of magnetic behavior described in a mean field approximation as an assembly of ferromagnetically coupled pentanuclear $Cu^{II}_3W^{V_2}$ units with very weak antiferromagnetic interactions between these units. Further work is being directed toward the construction of multifunctional architectures using 3d metal analogues of { $[Cu^{II}(\mu-4,4'-bpy)]^{2+}_n$ and related ligands in sequential self-assembly processes. We hope that the new class of magnetic multinuclear metal complexes revealed here represents a step toward supramolecular spintronics.⁸

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Supporting Information Available: Crystallographic information file (CIF), Figure S1 (asymmetric unit of 1), Figure S2 (H-bonded water molecules within the layer of 1), Scheme S1 (schematic structures of the W–CN–Cu linkages), and Scheme S2 (schemes of magnetic orbitals in 1). This material is available free of charge via the Internet at http://pubs.acs.org.

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